FEBS 14177

Isolation and characterization of a high molecular weight cytochrome from the sulfate reducing bacterium *Desulfovibrio gigas*

Liang Chena, Manuela M. Pereirab, Miguel Teixeirab, António V. Xavierb, Jean Le Galla

^aDepartment of Biochemistry, The University of Georgia, Athens, GA 30602, USA

bInstituto de Tecnologia Química e Biológica and Faculdade de Ciências e Tecnologia, Universidade Nova de Lisboa, Apt. 127, 2780 Oeiras, Portugal

Received 22 April 1994; revised version received 25 May 1994

Abstract

A high molecular weight c-type cytochrome (Hmc) was purified and characterized from Desulfovibrio gigas. The molecular weight was estimated to be 67 kDa by SDS-PAGE and its N-terminus is homologous to those of the 16 hemes containing high molecular weight cytochrome c from Desulfovibrio vulgaris strains Hildenborough and Miyazaki. The purified hemoprotein shows c-type cytochrome absorption spectrum with ε_{553} (red) = 368 mM⁻¹·cm⁻¹. A band at 640 nm, characteristic of high-spin hemes, was detected. The EPR spectra show the presence of two high-spin heme species, plus several non-equivalent low-spin hemes. The heme reduction potentials, at pH 7.6, range from -50 mV to -315 mV. In contrast to what has been described for D. vulgaris Hmc, the protein isolated from D. gigas directly accepts electrons from hydrogenase and further reduces other redox proteins.

Key words: Cytochrome c; Multiheme cytochrome; Desulfovibrio gigas; N-Terminal sequence; Electron transfer; Electron paramagnetic resonance

1. Introduction

Bacteria belonging to the *Desulfovibrio* genus are sulfate-reducers that can use sulfate or sulfite as respiratory substrates in the presence of various carbon sources. Several types of cytochromes of the b or c types have been isolated from them. D. gigas contains at least five different cytochromes: cytochrome b [1], tetraheme cytochrome c_3 [2], octaheme cytochrome c_3 [3], a partially identified membrane-bound cytochrome c [4], and a new type of flavohemo protein, a rubredoxin-oxygen oxidoreductase [5].

A high-molecular weight cytochrome c (Hmc), containing 16 hemes per molecule was purified and characterized from D. vulgaris strains Hildenborough [6–9] and Miyazaki [10,11]. Although Pollock et al. [6] failed to detect any cross-reaction in D. gigas with a DNA probe of D. vulgaris Hildenborough Hmc, we show in this report that a similar hemoprotein is present in D. gigas and describe its purification and characterization.

2. Materials and methods

2.1. Purification of Hmc

D. gigas (ATCC 19364) cells and soluble extract were obtained as in [12]. All purification procedures were performed at pH 7.6 and 4°C. The purity of Hmc was determined by the ratio of A_{553} (red)/ A_{280} (ox) and

Abbreviations: DEAE, diethyl aminoethyl; Hmc, high molecular weight c-type cytochrome; SDS-PAGE, sodium dodecylsulfate-polyacrylamide gel electrophoresis; Tris, tris (hydroxymethyl) aminomethane.

SDS-PAGE analysis. The dialyzed soluble extract from 1 kg of D. gigas frozen cells was loaded onto a DEAE-cellulose column (10 × 40 cm) equilibrated with 10 mM Tris-HCl buffer (buffer I). After loading, the column was washed with 11 buffer I. A linear Tris-HCl buffer gradient (0.01-0.5 M) of a total volume of 81 was applied. A fraction containing cytochromes was collected at 0.15 M, which was then applied to an hydroxyapatite column $(4.5 \times 30 \text{ cm})$ equilibrated with buffer I. An ascending linear gradient (0.001-0.15 M) of potassium phosphate buffer (pH 7.6) was applied with a total volume of 21. The eluted Hmc (at ~30 mM phosphate buffer) was dialyzed against water overnight and then loaded onto a DEAE-Biogel column (4 × 30 cm) equilibrated with buffer I. Hmc was eluted at 60-80 mM after application of a linear gradient of Tris-HCl buffer (0.01-0.2 M) with a total volume of 1.5 l. This partially purified Hmc was subjected to an electroelution procedure. The partially purified Hmc was loaded on a 7.5% PAGE gel and submitted to electrophoresis. The red Hmc band was then cut and loaded on to an electroelution chamber (V = 100 mV) and eluted with 50 mM NH₄HCO₃ overnight. After this step, the purified Hmc had an $A_{553}(\text{red})/A_{280}(\text{ox})$ ratio of 2.1. The yield of Hmc is 10 mg/kg frozen

2.2. Analytical methods

The protein concentration of Hmc was measured by the Lowry assay [13] with pure D. gigas tetraheme cytochrome c_3 as a standard. The molecular mass of Hmc was determined by 12.5% SDS-PAGE [14]. The heme content was determined from pyridine-ferrohemochrome spectrum, using the extinction coefficient of 29.1 mM $^{-1}$ ·cm $^{-1}$ at 550 nm from horse heart cytochrome c [15]. Total iron was measured by plasma emission spectroscopy using a Jarrell-Ash Model 750 atom comp spectrometer. UV/visible spectra were obtained using a Shimadzu spectrophotometer (UV 265). N-Terminal sequence was determined by the method of Edman and Begg [16] with an Applied Biosystem sequenator (Model 470A). EPR spectra were obtained as in [17]. Parallel-mode spectra were obtained with a Bruker ER 4116 Dual Mode cavity. Visible redox titrations were performed as described in [18].

3. Results and discussion

3.1. Chemical characterization

The purified Hmc shows a single band (67 kDa) on 12.5% SDS-PAGE. Although the pyridine-ferrohemo-

^{*}Corresponding author. Fax: (351) (1) 442 8766.

	1	5	10	15	20	25	30	
	1	1	1	1	į	1	1	
$\underline{\mathbf{D}} \cdot \underline{\mathbf{v}} \cdot \mathbf{Hmc}$	K-A-L	-P-E-G-P-	G-E-K-R-A-I	-L-I-E-I-	G-A-M-E-R-	F-G-K-L-D-	L-P-K-V-A	-F-R-
D.g.Hmc	A-T-P-G	-P-A-S-T-	A-E-P-K-V-I	-L-I-V-G-	E-L-A-V-G-	F-G-K-L-E-	M-P-F-V-M	-F-D
D.v.L-peptide	P-L	-P-G-A-T-	G-E-O-R-A-I	• • -L-V-E-I-	G-V-M-A-K-	F-T-N-L-E-	L-P-K-V	

Fig. 1. Alignment of N-terminal sequences of Hmc isolated from *D. gigas* and *D. vulgaris* strains Hildenborough [6] and Miyazaki [11]. *D.v.* Hmc, *D. vulgaris* Hildenborough Hmc; *D.g.* Hmc, *D. gigas* Hmc; *D.v.* L-peptide, *D. vulgaris* Miyazaki L-peptide of Hmc. A colon indicates two aligned residues are identical; a dot indicates that two aligned residues are similar. The alignment was performed using PC/Gene software package, version 6.7 [29].

chrome spectrum indicated that D. gigas Hmc contains 16 hemes of the c-type, only 13.4 atoms of iron per molecule were found by plasma emission. The lower iron content of D. gigas Hmc measured by plasma emission could have resulted from technical inaccuracy; however, a definitive heme content will only be established after full amino acid sequencing. It is to be noted that D. vulgaris Miyazaki Hmc was reported to have 11 hemes [10], a value that was suggested to be due to its high oxygen sensitivity [8]. The N-terminal sequence of D. gigas Hmc (Fig. 1) has 31.4% identity plus 8.6% similarity to D. vulgaris Hildenborough Hmc, and 28.6% identity plus 14.3% similarity to D. vulgaris Miyazaki Hmc L-peptide. D. vulgaris Hildenborough Hmc was suggested to be a periplasmic protein because a signal peptide was found in the sequence of the gene [6]. In contrast, the same Hmc was isolated from D. vulgaris Hildenborough cytoplasmic fractions (our unpublished result).

3.2. Electronic spectra

The UV/visible spectrum of oxidized Hmc shows peaks at 280 nm, 409 nm (Soret, $\varepsilon = 1616 \text{ mM}^{-1} \cdot \text{cm}^{-1}$), 529 nm and ~640 nm (Fig. 2), indicating the presence of high- and low-spin hemes. In the reduced form the Soret band shifts to 419 nm (ε = 2470 mM⁻¹·cm⁻¹) and the α and β peaks appear at 523 ($\varepsilon = 201 \text{ mM}^{-1} \cdot \text{cm}^{-1}$) and 553 nm ($\varepsilon = 368 \text{ mM}^{-1} \cdot \text{cm}^{-1}$), respectively (Fig. 2). A shoulder at ~429 nm, typical of high-spin ferrohemes [19], is also observed upon reduction. Even in highly concentrated samples (up to 100 mM), no peak is observed at the wavelength of 695 nm, characteristic of methionine bound hemes [19]. By using the usual molar absortivities for high- and low-spin ferri-hemes [19], the contribution at 640 nm accounts for approximately 10% of the the total heme content. The extinction coefficient of reduced D. gigas Hmc at 553 nm is approximately 4 times that of tetraheme cytochrome c_3 (ε_{553} (red) = 109 mM⁻¹·cm⁻¹) from the same organism, indicating that the new protein may contain 16 hemes. Degigas Hmc has a very low absorption at 280 nm, in agreement with its high heme content and suggesting that relatively few aromatic amino acid residues such as tryptophan are present.

3.3. EPR spectroscopy

The EPR spectra of oxidized D. gigas Hmc (Fig. 3) show clearly two high-spin components, with g_{max} at 6.54 and 6.1, corresponding to species with slightly different rhombic distortions ($E/D \sim 0.02$ and ~ 0.005 , respectively, see Fig. 3D,E). These two species are present in a 1:1 ratio in the pH range 6–11. Several low-spin components are also observed, with g_{max} values spreading from 3.45 (particularly well observed in the low temperature spectrum, Fig. 3A) to 2.93. Since a complete deconvolution of each species is not possible at present, the low-spin components were simulated in order to obtain an overall representation of the total spectral envelop, which could be used for the spectral integration (see below and Fig.

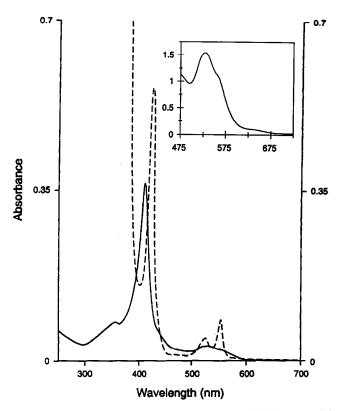


Fig. 2. The UV/visible spectra of *D. gigas* Hmc. Solid line, 0.24 μ M oxidized Hmc. Dash line, 0.24 μ M Hmc reduced by dithionite. Inserted curve, 12 μ M oxidized Hmc in 100 mM Tris buffer (pH 7.6).

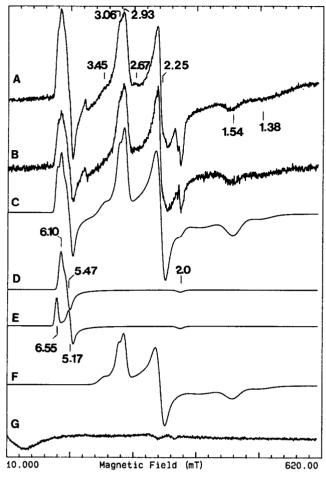


Fig. 3. EPR spectra of *D. gigas* Hmc, oxidized (A,B) and H₂/hydrogenase reduced (G). The full lines correspond to the theoretical simulations, generated using the g-values presented below, for the total spectrum (C), the high-spin species (D,E) and the low-spin species (F). Temperature, 4.2 K (A,G) and 19 K (B). Modulation amplitude, 0.9 mT; Microwave power, 2.4 mW; Microwave frequency, 9.63 GHz (A,B) and 9.38 GHz (G, parallel mode spectrum). The following g-values were used for the simulation: (i) for the high-spin species, 6.55, 5.17, 2.0 and 6.10, 5.47, 2.0; for the low-spin species, 2.93, 2.26, 1.54 (I), 3.06, 2.25,1.38 (II), 3.16, 2.23,1.0 (III) and 3.45, 1.90, 0.7 (IV). The simulations were added in the ratio 1:1:5:2.5:2.5:4 (see text).

3). The crystal field parameters for the species I-III are within the range found for hemes with a bis-histidine or histidine-methionine axial coordination [17,20] with a rhombic crystal field symmetry. The components with $g_{\text{max}} = 3.45$ have an almost axial ligand field, which, for bis-histidine ligation, indicates an almost perpendicularity of the imidazole planes. In some samples, a species with g_{max} at ~2.67 is observed, accounting for less than 0.2% of the total low-spin heme intensity. This species, also observed in *D. vulgaris* Hmc [8], has a g_{max} value in the range of hydroxide bound hemes [20], but does not increase in intensity upon raising the pH. The proportion of high- to low-spin hemes was determined by using the theoretical simulations of both types of species, adjusted

to the experimental spectrum obtained under non-saturating conditions (Fig. 3B). Assuming a zero-field splitting of 10 cm^{-1} for the high-spin hemes [21], the percentage of high- to low-spin hemes was calculated to be approximately 13, which suggests the presence of two high-spin hemes. The overall spectrum of D. gigas Hmc is similar to that of D. vulgaris Hildenborough Hmc: a closer look at the EPR spectrum of this last protein [8] shows the presence of two distinct high-spin species, apparently identical to those observed for the D. gigas protein. The authors referred to this species but did not include it in the simulation, leading to an underestimation of the total high-spin heme content.

D. gigas [NiFe]-hydrogenase can reduce all the Hmc hemes. The EPR spectra of reduced Hmc show only a broad signal at very low magnetic field, particularly well observed using paralell mode EPR (Fig. 3G), indicating the presence of high-spin ferro-hemes [22].

3.4. Redox potentials

The reduction potentials for *D. gigas* Hmc at pH 7.6 were estimated by redox titrations, measuring visible absorbance changes at 419, 429 and 553 nm as a function of the solution redox potential (Fig. 4). The experimental curves were adjusted with 4 Nernst equations, with reduction potentials of -80, -215 -255 and -315 mV, added in the ratio 4:4:4:2. The high-spin heme reduction potentials were estimated to be of about -50 mV, from the increase in absorbance at 429 nm (data not shown).

3.5. Electron transfer properties

It was reported that D. vulgaris Miyazaki Hmc could not be reduced by H_2 /hydrogenase unless in the presence of cytochrome c_3 [10]. As shown above, this is not the case for D. gigas Hmc (see Fig. 3G). Furthermore, under H_2 , the hydrogenase/Hmc mixture can induce the rapid reduction of D. gigas rubredoxin, which in turn fully reduces D. gigas rubredoxin-oxygen oxidoreductase [5] (data not shown). This last reaction is not possible with the tetraheme cytochrome c_3 .

4. Conclusions

The analysis of the *D. vulgaris* Hildenborough Hmc protein sequence deduced from molecular cloning [6] suggested that the heme binding sites and axial ligands are arranged as three complete and one incomplete tetraheme cytochrome c_3 -like domains. Actually, the array of the heme binding sites and potential sixth ligand His residues for this so-called incomplete domain is homologous to that of *Desulforomonas acetoxidans* triheme cytochrome $c_{551.5}$ (also known as cytochrome c_7) [23]. Since there is at least one high-spin heme in both redox states of *D. gigas* Hmc and no spectroscopic evidence for a

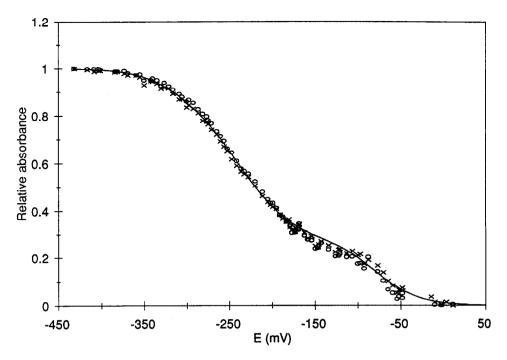


Fig. 4. Redox titration curve of *D. gigas* Hmc, obtained by following the absorption changes at 553 nm (0) and 419 nm (x). The solid line was obtained as described in the text.

methionine-bound heme was found, there is no need to invoke this type of axial ligation [6,8]. Thus, the spectroscopic and chemical data of D. gigas Hmc, and the sequence studies of the homologous protein from D. vulgaris taken together point to the presence of 14-15 bishistidinyl coordinated low-spin hemes and 1-2 high-spin hemes in D. gigas Hmc. Rossi et al. [9] have suggested that each of the Hmc domains interacts with one of the three hydrogenases found in D. vulgaris [24,25]. However, in the same paper, the authors present a scheme in which the [Fe]-hydrogenase of D. vulgaris is presented as the only hydrogenase reacting with Hmc, which is proposed to be part of a transmembrane electron transfer complex. As far as D. gigas Hmc is concerned, such schemes do not seem to apply since this strain is known to contain only one hydrogenase [26]. However, the possibility still exists that other hydrogenases different from the classical [Fe]-, [NiFe]- and [NiFeSe]-containing enzymes have remained undetected. The electron transfer properties of D. gigas Hmc show that, like the tetraheme cytochrome c_3 , it can function as an electron acceptor of hydrogenase [27] or as an electron donor to rubredoxin, but that it may play a physiological role other than in D. vulgaris because the latter organism does not seem to contain an oxygen-linked rubredoxin-oxygen oxidoreductase (see above).

Our finding that the N-terminus of *D. gigas* Hmc is somewhat different from those of the *D. vulgaris* Hmc's can explain the results of Pollock et al. [6], who failed to detect any cross-reaction in *D. gigas* with a DNA probe

of *D. vulgaris* Hildenborough Hmc. As noted by Voordouw [28], this negative result is not a proof that a similar hemoprotein is absent in *D. gigas*.

Acknowledgements: We thank Mrs. M. Regalla for amino acid sequence determinations and the personnel of The UGA Fermentation Plant for growing the bacterial cells. This work was supported by grants of NIH (GM 41482) and NSF (DMB 9005734), to J.L, and by JNICT grant PMCT-CEN-649/93 to MT.

References

- Hatchikian, E.C. and LeGall, J. (1972) Biochim. Biophys. Acta 267, 479-484.
- [2] Le Gall, J., Mazza, G. and Dragoni, N. (1965) Biochim. Biophys. Acta 99, 385-387.
- [3] Hatchikian, E.C., Bruschi, M., Le Gall, J. and Dubourdieu, M. (1969) Bull. Soc. Fr. Physiol. Veg. 15, 381-390.
- [4] Chen, L., Liu, M.-Y. and Le Gall, J.(1993) Arch. Biochem. Biophys. 303, 44-50.
- [5] Chen, L., Liu, M.-Y., Le Gall, J., Fareleira, P., Santos, H. and Xavier, A.V. (1993) Biochem. Biophys. Res. Commun. 193, 100– 105
- [6] Pollock, W.B.R., Loutfi, M., Bruschi, M., Rapp-Giles, B.J., Wall, J.D. and Voordouw, G. (1991) J. Bacteriol. 173, 220-228.
- [7] Higuchi, Y., Inaka, K., Yasuoka, N. and Yagi, T. (1987) Biochim. Biophys. Acta 911, 341–348.
- [8] Bruschi, M., Bertrand, P., More, C., Leroy, G., Bonicel, J., Haladjian, J., Chottard, G., Pollock, W.B.R. and Voordouw, G. (1992) Biochemistry 31, 3281-3288.
- [9] Rossi, M., Pollock, W.B.R., Reij, M.W., Kevn, R.G., Fu, R. and Voordouw, G. (1993) J. Bacteriol. 175, 4699-4711.
- [10] Yagi, T. and Ogata, M. (1990) in: Proceedings of the FEMS Symposium on Microbiology and Biochemistry of Strict Anaerobes

- Involved in Interspecies Hydrogen Transfer (Belaich, J.-P., Bruschi, M. and Garcia, J.-L., eds) pp. 227-249, Plenum Press, New York.
- [11] Tasake, C., Ogata, M., Yagi, T. and Tsugita, A. (1991) Prot. Seq. Data Anal. 4, 25-27.
- [12] Hatchikian, E.C., Le Gall, J., Bruschi, M. and Dubourdieu, M. (1972) Biochim. Biophys. Acta 258, 701-708.
- [13] Lowry, O.H., Rosebrough, N.J., Farr, A.L. and Ranadall, R.J. (1951) J. Biol. Chem. 193, 265-275.
- [14] Laemmli, U.K. (1970) Nature 227, 680-685.
- [15] Fuhrlop, J.-H, and Smith, K.M. (1975), in: Porphyrins and Metalloporphyrins, (Smith, K.E., ed.) pp. 757-861, Elsevier, Amsterdam.
- [16] Edman, P. and Begg, G. (1976) Eur. J. Biochem. 1, 80-91.
- [17] Teixeira, M., Campos, A.P., Aguiar, A.P., Costa, H.S., Santos, H., Turner, D.L. and Xavier, A.-V., (1993) FEBS Lett. 317, 233-236.
- [18] Costa, H.S., Santos, H., Turner, D.L. and Xavier, A.X. (1992) Eur. J. Biochem. 208, 427-433.
- [19] Moore, G.R. and Pettigrew, G.W. (1990) in: Cytochrome c: Evolutionary, Structural and Physiological Aspects, Springer-Verlag, Reglin
- [20] Blumberg, W.E. and Peisach, J. (1971) in: Structure and Function

- of Macromolecules and Membranes, vol. 2 (Chance, B., Yonetani, T. and Mildvan, A., eds.) pp. 215–229, Academic Press, NY.
- [21] Aasa, R., Albracht, S.P.J., Falk, K.-E., Laane, B. and Vänngard, T. (1976) Biochim. Biophys. Acta 422, 260-272.
- [22] Hendrich, M.P. and Debrunner, P.G. (1988) J. Magn. Res. 78, 133-141.
- [23] Ambler, R.P. (1971) FEBS Lett. 18, 351-353.
- [24] Lissolo, T., Choi, E.S., LeGall, J. and Peck Jr., H.D. (1986) Biochem. Biophys. Res. Commun. 204, 247-250.
- [25] Prickril, B.C., He, S.H., Li, C., Menon, N., Choi, E.-S., Przybyla, A.E., Der Vartanian, D.V., Peck Jr., H.D., Fauque, G., LeGall, J., Teixeira, M., Moura, I., Moura, J.J.G., Patil, D. and Huynh, B.H. (1987) Biochem. Biophys. Res. Commun. 149, 369-377.
- [26] Voordouw, G., Niviere, V., Ferris, F.G., Fedorak, P.M. and Westlake, D.W.S. (1990) Appl. Environ. Microbiol. 56, 3748-3754.
- [27] Bell, G.R., Lee, J.-P, Peck Jr., H.D. and LeGall, J. (1978) Biochimie 60, 315-320.
- [28] Voordouw, G. (1992) in: The Sulfate-reducing Bacteria: Contemporary Perpectives (Odom, J.M. and Singleton Jr., R. eds.) ch. 5, Springer-Verlag, NY.
- [29] PC-Gene software package, Intelligenetics Inc. (1992), 700 East El Camino Real, Mountain View, CA, USA.